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Line radiative transfer and statistical equilibrium*

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Abstract. Atomic and molecular line emission from protoplanetary disks contains key information of their detailed physical and chemical structures. To unravel those structures, we need to understand line radiative transfer in dusty media and the statistical equilibrium, especially of molecules. I describe here the basic principles of statistical equilibrium and illustrate them through the two-level atom. In a second part, the fundamentals of line radiative transfer are introduced along with the various broadening mechanisms. I explain general solution methods with their drawbacks and also specific difficulties encountered in solving the line radiative transfer equation in disks (e.g. velocity gradients). I am closing with a few special cases of line emission from disks: Radiative pumping, masers and resonance scattering.

1 Introduction

The previous chapter “Line Observations in Disks” (Dionatos 2015) introduced the different degrees of freedom a molecule has (rotation, vibration, electronic levels) and the respective nomenclature for atomic and molecular lines. Spectra of molecules are particularly powerful for deducing physical conditions in the observed regions, e.g. also in protoplanetary disks. The role of various molecules and specific types of transitions is nicely summarized in Table 5.1 of Stahler & Palla (2004). For disks, interesting diagnostics are CS and CN, which are high density probes, and thus are — especially in their higher rotational lines — less contaminated than CO by emission from the surrounding cloud.

A rule of thumb to remember is that lines preferentially originate close to where the critical density is reached and also close to where the optical depth of the line becomes one. The critical density n_{cr} is defined as the ratio between the Einstein coefficient for spontaneous emission A_{ul} and the collision rate between the upper and lower level of the line transition C_{ul}

$$n_{\text{cr}} = \frac{A_{ul}}{C_{ul}} . \quad (1)$$

If the density is higher than the critical density, the level populations will follow the Boltzmann distribution due to efficient collisional coupling and the line will be in Local Thermodynamic Equilibrium (LTE).

The interplay between critical density and excitation temperature in defining the optimum emitting conditions for various types of transitions and molecules is shown in figure 1 in the chapter by Dionatos (2015). It nicely illustrates that the various types of CO lines ranging from low rotational

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lines to high rotational lines and then ro-vibrational transitions have the potential to probe the entire disk from the low density cold outer regions to the very hot, dense inner regions (see Fig. 1).

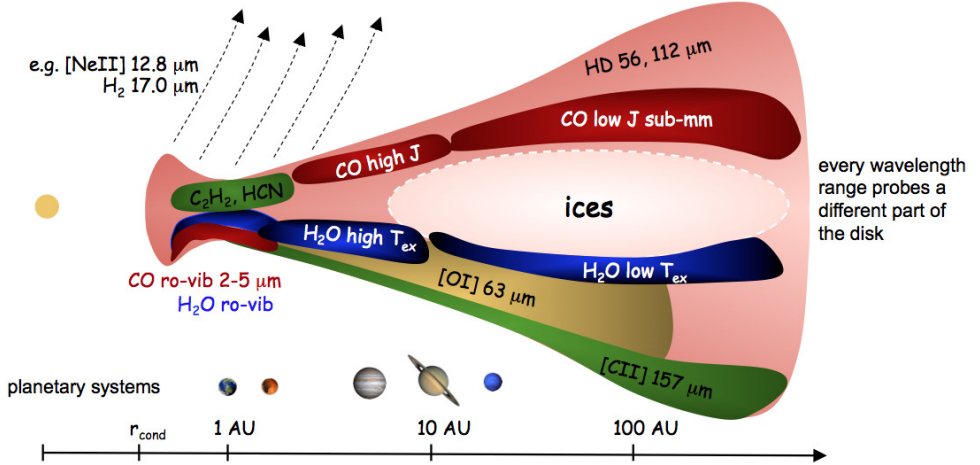


Figure 1. Disk sketch with the various molecular and atomic line emitting regions.

2 Statistical Equilibrium

Line fluxes depend on the level populations, or more directly on the column density of an atom/molecule in the upper energy level. We will simplify the radiative transfer for a moment and only consider the probability, β , that a line photon, emitted somewhere in a column of gas that we study, escapes the medium. The escape probability depends on the optical depth τ_{ul} at line center frequency ν_{ul} . We will explain that concept later in more detail (see also chapter by Woitke 2015) and also get back to the full radiative transfer. Then, we can write the relation between the column density of the upper level, N_u , and the line flux F_{ul} as

$$F_{ul} \approx N_u A_{ul} h\nu_{ul} \beta(\tau_{ul}) \frac{\Omega_{\text{source}}}{4\pi}, \quad (2)$$

where A_{ul} is the Einstein coefficient for spontaneous emission and Ω_{source} the solid angle of the line emitting region. This equation directly illustrates the importance of determining the level populations to calculate the column densities and hence compute line fluxes. In the following, we will discuss how to calculate these level populations in LTE and non-LTE (statistical equilibrium, SE).

2.1 Local Thermodynamic Equilibrium

In the case of LTE, the level populations within a specific ionization state of an atom or molecule are given by the Boltzmann equation

$$\frac{n_u}{n_l} = \frac{g_u}{g_l} e^{-\Delta E/kT}, \quad (3)$$

where n_u and n_l are the level populations of the upper and lower energy level and g_u and g_l their statistical weights, respectively. $\Delta E = E_u - E_l$ is the energy difference between the two levels, k the Boltzmann constant and T is the gas temperature. Note that even if LTE does not hold, one can still define an excitation temperature T_{ex} that relates the two level populations through

$$\frac{n_u}{n_l} = \frac{g_u}{g_l} e^{-\Delta E/kT_{\text{ex}}} . \quad (4)$$

2.2 Ionization Equilibrium

The ionization balance of atoms and molecules in LTE follows from the Saha equation

$$\frac{n(X_{i+1})}{n(X_i)} = \frac{(2\pi mkT)^{3/2}}{h^3 n_e} \frac{2 Z(X_{i+1})}{Z(X_i)} \exp\left(-\frac{\chi_{i,i+1}}{kT}\right) , \quad (5)$$

where $n(X_i)$ is the level population in the ionization state X_i of the atom/molecule, n_e is the electron density, $\chi_{i,i+1}$ is the ionization energy from state X_i to X_{i+1} , m is the mass of the atom/molecule, and T is the gas temperature. The partition function $Z(X_i)$ of the ionization state X_i is defined as

$$Z(X_i) = \sum_j g_j e^{-E_j/kT} . \quad (6)$$

2.3 Non Local Thermodynamic Equilibrium

If the assumption of LTE breaks down, we have to solve the equations for non-LTE statistical equilibrium (SE). This means that we need to take into account all radiative and collisional processes that govern the distribution of level populations. The radiative processes are subject to the quantum mechanical selection rules and comprise spontaneous emission, stimulated emission and absorption, which depend on the respective Einstein coefficients. Collisional coupling can occur between any two levels; however, collision cross sections are generally larger between closely spaced energy levels and dipole allowed transitions. For each energy level, we can balance the rate for processes populating that level and those de-populating that same level. In equilibrium, $d\mathbf{n}/dt = 0$ holds, where \mathbf{n} is the vector containing all level populations n_i .

The complete set of equations for SE becomes then

$$\begin{aligned} \frac{dn_i}{dt} = & \sum_{j>i} n_j (A_{ji} + B_{ji}P(v_{ji})) + \sum_{j<i} n_j B_{ji}P(v_{ji}) + \sum_{j\neq i} n_j C_{ji} \\ & - n_i \sum_{j<i} (A_{ij} + B_{ij}P(v_{ij})) - n_i \sum_{j>i} B_{ij}P(v_{ij}) - n_i \sum_{j\neq i} C_{ij} . \end{aligned} \quad (7)$$

The first term sums over all emissions from higher levels ending up in level i , the second term over all absorptions from lower levels into level i . The third term adds up all collisional rates C_{ji} from lower and higher levels into level i . The fourth to sixth term add up the respective radiative and collisional depopulation of level i . $P(v_{ij})$ denotes the local mean intensity averaged over ray directions \vec{n} and local line profile functions $\phi_{ij}(v, \vec{n})$ as

$$P(v_{ij}) = \frac{1}{4\pi} \iint \phi_{ij}(v, \vec{n}) I_v(\vec{n}) dv d\Omega , \quad (8)$$

which can cause absorption and stimulated emission. $I_\nu(\vec{n})$ is the spectral intensity at frequency ν in direction \vec{n} . The three different Einstein coefficients for spontaneous emission A_{ji} , stimulated emission B_{ji} , and absorption B_{ij} are coupled through the following relations

$$B_{ji} = \frac{c^2}{2h\nu^3} A_{ji} \quad (9)$$

and

$$g_i B_{ij} = g_j B_{ji} . \quad (10)$$

Note that the exact form of these relations (first term of constants) depends on the units in which the radiation field is given.

This set of equations can be written in vector form as

$$\mathbf{M}\mathbf{n} = \mathbf{b} , \quad (11)$$

where \mathbf{n} is an N -dimensional vector composed of the individual level populations n_i for all N levels and \mathbf{M} is an $N \times N$ matrix composed of elements M_{ij} . For example, the column u with $u > i$ contains the elements

$$M_{iu} = (A_{ui} + B_{ui}P(\nu_{ui}) + C_{ui}) . \quad (12)$$

The largest amount of work lies in the compilation of all the atomic/molecular data, i.e. the level energies, statistical weights, radiative transitions, Einstein A coefficients, collision cross sections. Several databases provide this data such as the LAMDA database (Schöier et al. 2005), NIST, and CHIANTI (Dere et al. 1997).

2.4 Solving the Statistical Equilibrium

Once all radiative and collisional data is compiled for a specific atom/molecule, Eq. (11) can be inverted and solved for \mathbf{n} . We need to respect, however, a very important boundary condition, namely the particle conservation that ensures that the sum of all level populations n_i adds up to the total density of that atom/molecule n_{tot} . The way to impose this implicitly in the solution of the SE equation is to replace one of the equations in Eq. (11) with the conservation equation

$$\sum_i n_i = n_{\text{tot}} . \quad (13)$$

The matrix \mathbf{M} then becomes

$$\mathbf{M} = \begin{bmatrix} M_{11} & M_{12} & \cdots & M_{1N-1} & M_{1N} \\ M_{21} & M_{22} & \cdots & M_{2N-1} & M_{2N} \\ \vdots & \vdots & \ddots & \vdots & \vdots \\ M_{N-11} & M_{N-12} & \cdots & M_{N-1N-1} & M_{N-1N} \\ 1 & 1 & \cdots & 1 & 1 \end{bmatrix} \quad \mathbf{b} = \begin{bmatrix} 0 \\ 0 \\ \vdots \\ 0 \\ n_{\text{tot}} \end{bmatrix} . \quad (14)$$

This matrix equation can be solved using standard numerical recipes to invert the matrix such as LU decomposition. The problem scales with N^3 , making evident why often limited atoms/molecules are used, i.e. the amount of ionization stages and levels being restricted according to the physical conditions in the region to be investigated. An example is the restriction to pure rotational energy levels of the ground electronic and ground vibrational level of CO in the case of molecular clouds, where densities and temperatures are low ($n \sim 10^4 \text{ cm}^{-3}$, $T \sim 10 \text{ K}$).

2.5 The Two-level Atom

To provide better insight into the SE, we will in the following consider the two-level atom as an example. The atom has two energy levels with populations n_0 and n_1 , which are connected through a single line with frequency ν_{10} that has an Einstein coefficient A_{10} . The SE in Eq. (7) becomes in that case

$$\frac{dn_1}{dt} = n_0 B_{01} P(\nu_{10}) + n_0 C_{01} - n_1 [A_{10} + B_{10} P(\nu_{10})] - n_1 C_{10} = 0 \quad (15)$$

We can re-arrange this equation to isolate the ratio between the two level populations — making also use of the relations between the Einstein A and B coefficient —

$$\frac{n_1}{n_0} = \frac{A_{10} \frac{c^2}{2h\nu_{10}^3} \frac{g_1}{g_0} P(\nu_{10}) + C_{01}}{A_{10} + A_{10} \frac{c^2}{2h\nu_{10}^3} P(\nu_{10}) + C_{10}}. \quad (16)$$

We see immediately that in the absence of a radiation field, $P(\nu_{10})=0$, this simplifies to

$$\frac{n_1}{n_0} = \frac{C_{01}}{A_{10} + C_{10}}. \quad (17)$$

Using the relation between the upwards and downwards collision rates

$$\frac{C_{01}}{C_{10}} = \frac{g_1}{g_0} \exp\left(-\frac{E_{10}}{kT_{\text{gas}}}\right) \quad (18)$$

we find for very large collision rates ($C_{10} \gg A_{10}$) that this equation becomes the Boltzmann equation. Hence, in that case, the two level populations are in LTE according to the local gas temperature T_{gas} .

An example of a simple two level atom is the two fine structure levels of the ground state of ionized carbon: Lower level $^2P_{3/2}$ and upper level $^2P_{1/2}$. They are separated by an energy E_{10} of 92 K and the emission line has a wavelength of $157.7 \mu\text{m}$. Since the critical density of this line is very low, it is widely observed in the interstellar medium (ISM, e.g. Pineda et al. 2013).

A second case to be considered is a situation where the collision rates are much smaller than the radiative rates ($C_{10} \rightarrow 0$ and $C_{01} \rightarrow 0$). In that case, Eq. (16) becomes

$$\frac{n_1}{n_0} = \frac{g_1}{g_0} \frac{\frac{c^2}{2h\nu_{10}^3} P(\nu_{10})}{1 + \frac{c^2}{2h\nu_{10}^3} P(\nu_{10})}. \quad (19)$$

If $P(\nu_{10})$ is a blackbody radiation field of temperature T_{gas} , this becomes

$$\frac{n_1}{n_0} = \frac{g_1}{g_0} \frac{\frac{c^2}{2h\nu_{10}^3} \left(\frac{2h\nu_{10}^3}{c^2} \frac{1}{e^{h\nu_{10}/kT_{\text{gas}}} - 1} \right)}{1 + \frac{c^2}{2h\nu_{10}^3} \left(\frac{2h\nu_{10}^3}{c^2} \frac{1}{e^{h\nu_{10}/kT_{\text{gas}}} - 1} \right)} = \frac{g_1}{g_0} \exp\left(-\frac{E_{10}}{kT_{\text{gas}}}\right) \quad (20)$$

Note, that we never imposed LTE in this case, on the contrary, we started with negligible collision rates. Hence, radiation can also impose LTE on the level populations provided that the radiation field is that of a blackbody. Of course, this will be a valid assumption in the optically thick regions of the disk.

3 Radiative Transfer

Now that we understand how the different rotational, vibrational and electronic levels of an atom/molecule are populated, we can address the second part of understanding line emission: how does radiation cross the medium to reach the observer. From the equations of SE it became already evident that the background radiation field can pump level populations under certain conditions. Hence, the two problems of SE and radiative transfer are closely intertwined, making the combined problem very difficult to solve even numerically. We will come back to that later when we address simplifications made in disk research. The next few paragraphs will first illustrate the basic theory of line emission/absorption and line broadening.

3.1 Line Emission Coefficient

The line emission coefficient ϵ_v^{ij} is related to the Einstein A coefficient, the level population of the upper level n_i and the line profile function $\phi_{ij}(\nu)$

$$\epsilon_v^{ij} = \frac{h\nu_{ij}}{4\pi} n_i A_{ij} \phi_{ij}(\nu) . \quad (21)$$

The line profile function describes how a line at central frequency ν_{ij} is broadened due to the uncertainty principle, collisions, thermal and turbulent motions in the gas, see Sect. 3.3. The line can also be Doppler-shifted by systematic motions of the gas in the observers frame, which causes the line profile function to become direction dependent $\phi_{ij} = \phi_{ij}(\nu, \vec{n})$ which is, however, not considered in this section. The line profile itself is normalized to one, $\int \phi_{ij}(\nu) d\nu = 1$, with the effect that the total line emission is smeared out over a certain frequency range symmetric around the rest frequency of the line ν_{ij} . A useful characterization of a line profile — also in observations — is the Full Width Half Maximum (FWHM), which measures the width of the profile at half its peak value.

3.2 Line Absorption Coefficient

The line absorption coefficient α_v^{ij} has two components, stimulated emission B_{ij} and absorption B_{ji} . In that case, the sign of the absorption coefficient immediately hints to the presence of laser/maser¹ emission, which will be discussed at the end of this chapter. The line absorption coefficient can be expressed as

$$\alpha_v^{ij} = \frac{h\nu_{ij}}{4\pi} (n_j B_{ji} - n_i B_{ij}) \phi_{ij}(\nu) , \quad (22)$$

where n_i and n_j are the upper and lower level populations respectively and $\phi_{ij}(\nu)$ is again the line profile function.

3.3 Line Broadening

The total line profile is a convolution of profiles from various types of broadening mechanisms: natural broadening, collisional broadening, thermal and turbulent broadening. The first two are related to the lifetime of the upper level of the transition and lead to a Lorentzian profile, while the second two cause Doppler shifts of the frequency and hence produce Gaussian profiles. The convolution of a Lorentzian and a Gaussian profile is a Voigt profile.

¹Laser stands for Light Amplification by Stimulated Emission of Radiation and maser for Microwave Amplification by Stimulated Emission of Radiation.

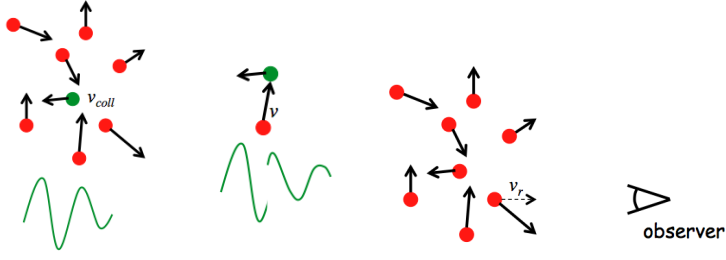


Figure 2. Sketch of natural broadening (left), collisional broadening (middle) and thermal broadening (right).

Natural Broadening

Natural broadening is caused by the uncertainty principle. The shorter the lifetime of a state, the larger the energy uncertainty of the state

$$\Delta E \Delta t \sim \hbar = \frac{h}{2\pi}, \quad (23)$$

and the natural line width thus becomes

$$\Delta \nu \sim \frac{\Delta E}{h} \sim \frac{1}{2\pi \Delta t}. \quad (24)$$

The width of the profile is related to the spontaneous de-excitation rate from level i into all other levels j , $\gamma = \sum_j A_{ij}$ (Fig. 2). With this, the Lorentzian line profile can be written as (indices i, j omitted)

$$\phi_{ij}(\nu) = \frac{\gamma/4\pi^2}{(\nu - \nu_{ij})^2 + (\gamma/4\pi)^2}. \quad (25)$$

Collisional Broadening

Atoms and molecules in a gas collide frequently and this reduces the lifetime in the different energy levels (Fig. 2). The broadening $\Delta \nu$ is proportional to the collision frequency ν_{coll} . Hence, collisional broadening becomes more important at higher densities. The line profile from the combined effect of natural and collisional broadening is still a Lorentzian

$$\phi_{ij}(\nu) = \frac{\Delta \nu_{\text{tot}}/4\pi^2}{(\nu - \nu_{ij})^2 + (\Delta \nu_{\text{tot}}/4\pi)^2}, \quad (26)$$

with the width $\Delta \nu_{\text{tot}}$ being the sum of the two widths ($\gamma + 2\nu_{\text{coll}}$).

In most astrophysical cases, natural broadening is negligible. Collisional broadening plays mostly a role at very high densities, such as the ones reached in stellar atmospheres $\gg 10^{12} \text{ cm}^{-3}$. This leaves thermal and turbulent broadening as the dominant mechanisms under ISM, molecular cloud and protoplanetary disk conditions.

Thermal Broadening

The thermal velocities of atoms and molecules show a random Gaussian 3D velocity distribution (Fig. 3), a Maxwell-Boltzmann distribution

$$f_v = 2\pi v^2 \sqrt{\left(\frac{m}{2\pi k T_{\text{gas}}}\right)^3} \exp\left(-\frac{mv^2}{2k T_{\text{gas}}}\right). \quad (27)$$

Here, f_v denotes the probability of finding an atom/molecule with a velocity close to v . The other quantities are the mass of the atom/molecule m and the gas temperature T_{gas} . Due to these random velocities, the frequency of the line ν_{ij} is shifted according to the projected radial velocity component v_r in the line of sight to an observer

$$\nu = \nu_{ij} \left(1 + \frac{v_r}{c} \right). \quad (28)$$

The resulting Doppler frequency shift is

$$\Delta\nu_D = \nu_{ij} \frac{v_r}{c}. \quad (29)$$

The most probable velocity of an atom/molecule of mass m in a gas of temperature T_{gas} is

$$v_r = v_{\text{therm}} = \sqrt{\frac{2kT_{\text{gas}}}{m}}. \quad (30)$$

A particle with this velocity has an energy of kT_{gas} . A simple expression to remember is the most probable velocity of a hydrogen atom

$$v_{\text{therm}}(\text{H}) = 1.3 \left(\frac{T_{\text{gas}}}{100 \text{ K}} \right)^{1/2} \text{ km/s}. \quad (31)$$

From this, one can easily scale for atoms/molecules of different mass or a gas of different temperature. The line profile resulting from thermal Doppler broadening is a Gaussian profile

$$\phi_{ij}(\nu) = \frac{1}{\Delta\nu_D \sqrt{\pi}} \exp \left(-\frac{(\nu - \nu_{ij})^2}{(\Delta\nu_D)^2} \right), \quad (32)$$

with the width $\Delta\nu_D$ (Eq. 29).

Turbulent Broadening

In the presence of turbulent motions v_{turb} , the width of the profile becomes

$$\Delta\nu_D = \frac{\nu_{ij}}{c} \left(\frac{2kT_{\text{gas}}}{m} + v_{\text{turb}}^2 \right)^{1/2}. \quad (33)$$

The root mean square of the thermal and turbulent velocities is often denoted by the parameter b in studies of the Interstellar Medium (ISM). Typical turbulent velocities for disks are inferred to be below 1 km/s.

The Gaussian profile is normalized to one. Its peak value at ν_{ij} and FWHM can be calculated from the width

$$\phi_{ij}(\nu) = \frac{1}{\Delta\nu_D \sqrt{\pi}}, \quad \text{FWHM} = 2 (\ln 2)^{1/2} \Delta\nu_D \approx 1.665 \Delta\nu_D. \quad (34)$$

Voigt profiles

A Voigt (V) profile is the convolution of a Lorentzian (L , width $\Delta\nu_{\text{tot}}$) and a Gaussian (G , width $\Delta\nu_D$) profile

$$V(\nu, \Delta\nu_D, \Delta\nu_{\text{tot}}) = \int_{-\infty}^{\infty} G(\nu', \Delta\nu_D) L(\nu - \nu', \Delta\nu_{\text{tot}}) d\nu'. \quad (35)$$

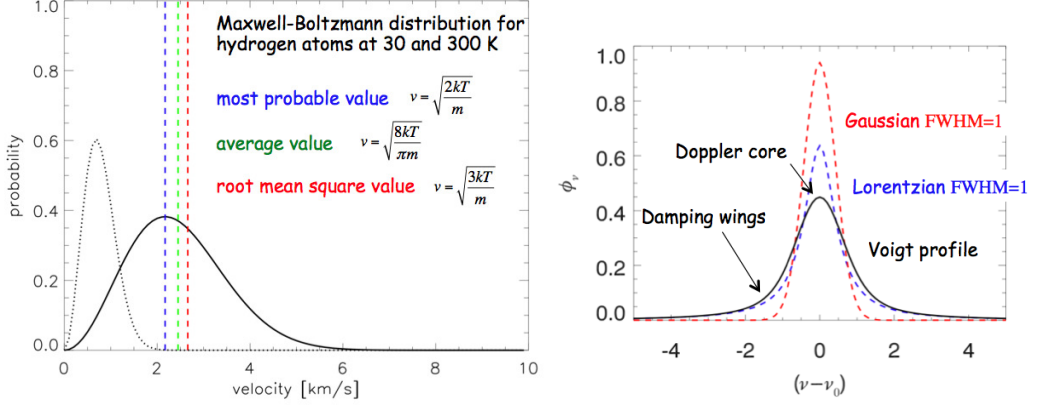


Figure 3. Left panel: Velocity distribution for hydrogen atoms at two different gas temperatures. Shown in color are the most probable, average and root mean square velocity of the distribution. Right panel: Comparison of a Lorentzian, Gaussian and Voigt profile.

It can be expressed as

$$\phi(\nu) = \frac{1}{\Delta\nu_D \sqrt{\pi}} H(\alpha, u), \quad (36)$$

with the Voigt function

$$H(\alpha, u) = \frac{\alpha}{\pi} \int_{-\infty}^{\infty} \frac{\exp(-y^2)}{\alpha^2 + (u - y)^2} dy \quad (37)$$

and the two parameters

$$\alpha = \frac{\Delta\nu_{\text{tot}}}{4\pi\Delta\nu_D}, \quad u = \frac{\nu - \nu_{ij}}{\Delta\nu_D}. \quad (38)$$

Fig. 3 (right panel) illustrates the shape of the Voigt profile in comparison with a Lorentzian and Gaussian, both having a FWHM of one and being normalized to one. The Voigt profile has wide damping wings (like the Lorentzian) and a Doppler core.

3.4 The Radiative Transfer Equation

The radiative transfer equation (RTE) describes how radiation is transported through a medium over a certain physical path length s

$$\frac{dI_\nu}{ds} = \alpha_\nu^{\text{ext}} (S_\nu - I_\nu). \quad (39)$$

The absorption coefficient in units of cm^{-1} is here the total extinction coefficient due to dust and line absorption $\alpha_\nu^{\text{ext}} = \alpha_\nu^{\text{dust}} + \alpha_\nu^{\text{line}} \phi_\nu$ with ϕ_ν being the line profile function. The source function is the ratio between the emission and absorption coefficients

$$S_\nu = \frac{\epsilon_\nu^{\text{dust}} + \epsilon_\nu^{\text{line}} \phi_\nu}{\alpha_\nu^{\text{dust}} + \alpha_\nu^{\text{line}} \phi_\nu}. \quad (40)$$

The dust emission coefficient ϵ_v^{dust} is composed of thermal emission and scattering, and can be calculated from the dust temperature T_{dust} and the local mean intensity J_v (assuming isotropic scattering)

$$\epsilon_v^{\text{dust}} = \alpha_v^{\text{dust,abs}} B_v(T_{\text{dust}}) + \alpha_v^{\text{dust,sca}} J_v, \quad (41)$$

where $\alpha_v^{\text{dust}} = \alpha_v^{\text{dust,abs}} + \alpha_v^{\text{dust,sca}}$. Calculating these two quantities with continuum radiative transfer and various numerical methods, such as Monte Carlo, has been discussed in the chapter by Pinte (2015). Using the definition of optical depth $d\tau_v = -\alpha_v^{\text{ext}} ds$, we can simplify the form of the RTE to

$$\frac{dI_v}{d\tau_v} = I_v - S_v. \quad (42)$$

The formal solution of this equation can be written as $J_v(\vec{r}) = \Lambda_v(\vec{r}) S_v$, where $\Lambda_v(\vec{r})$ is the famous Lambda integral operator. We can interpret this as a certain prescription operating on all source functions in the medium to obtain the frequency dependent radiation field J_v at position \vec{r} . If we imagine that we discretized our computational volume, the elements of the Λ operator couple each grid point with each other grid point in our volume. The coupling between SE and RT becomes evident in the source function, which contains the line emission and absorption coefficients, which in turn depend on the level populations (see Eq. 40).

3.5 Lambda Iteration

As stated above, the SE equations and the RTE are coupled. The source function depends on the radiation field itself, thus making the problem non-linear. Also, the level populations are local quantities and the radiation field is a global quantity. This means that local and global quantities are intrinsically coupled. Each iteration transports information roughly speaking over one mean free path of the photon. This makes an iterative solution scheme very slow in propagating the local information through the medium. In the presence of large velocity gradients, many frequency points will be needed, again slowing down the computation. Last but not least, it has been shown that the pure Lambda Iteration converges even before the solution is reached and it does not necessarily converge to the true solution. Hence, defining convergence criteria is very hard.

The full problem can be solved by inverting the Λ operator. However, in the case of velocity gradients or broad line profiles, i.e. frequency coupling in a moving medium or across lines, the matrix can be very large. Hence, an iterative approach has numerical advantages over a straight inversion. Seminal references that describe the concept of the Lambda Iteration and the discretization are Mihalas (1978) and Kalkofen (1984). The basic scheme of the iteration procedure is outlined in Fig. 4 (left panel).

At the beginning of the iteration, LTE and a blackbody radiation field corresponding to the local temperature can be used as an initial guess for the level populations n_i and the radiation field J_v allowing to compute the initial guess of the source function $S_v^{(0)}$. With this initial guess, the RTE can be formally solved for $J_v^{(1)}$. The radiation field is then used to solve the SE equations for the level populations n_i . From these, the line absorption and emission coefficients can be calculated and also the new source function $S_v^{(1)}$. This leads to the iterative scheme

$$J_v^{(n)} = \Lambda_v S_v^{(n-1)}. \quad (43)$$

Defining the Lambda operator explicitly is not required. Often the RTE is simply solved along pre-defined rays through the computation volume (e.g. methods of long or short characteristics). Hence, the mathematical scheme outlined above rather provides insight into the problem than serves as a numerical implementation guideline.

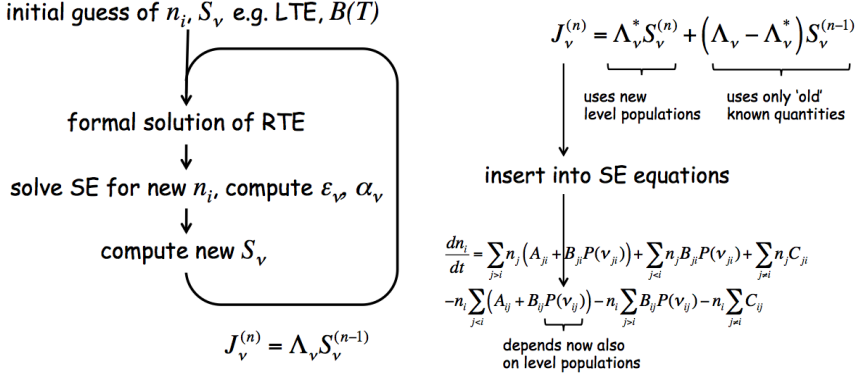


Figure 4. Left panel: Schematic flow of calculation for the Lambda Iteration (LI). Right panel: Improved concept of the Accelerated Lambda Iteration (ALI).

3.6 Accelerated Lambda Iteration

Having noted the convergence problems of the Lambda Iteration itself, we introduce now the more practical scheme of the Accelerated Lambda Iteration which in fact is much faster and does converge. Figure 4 (right panel) shows the basic idea of splitting the Lambda operator into two parts, an approximate Lambda operator Λ^* and the difference between the true and approximate Lambda operators $(\Lambda - \Lambda^*)$

$$J_v^{(n)} = \Lambda_v^* S_v^{(n)} + (\Lambda_v - \Lambda_v^*) S_v^{(n-1)}. \quad (44)$$

This method presents an advantage if the approximate operator can be constructed in such a way that it is much simpler than the true operator but at the same time contains the essential properties of the true operator. Details on constructing such approximate operators can be found in the review by Hubeny (2003) and references therein. This method of an approximate Lambda operator is often called the method of deferred corrections (Cannon 1973). This becomes clearer, if we study the new scheme in detail: The approximate operator acts on the new source function, which depends on the new level populations $n_i^{(n)}$. The second part of Eq. (44) makes only use old 'old' known quantities. The solution of the RTE is then inserted into the SE equations. The radiation field enters in the absorption and stimulated emission terms. Since J_v now depends on $n_i^{(n)}$, this makes the SE equations non-linear in the n_i 's and hence more difficult to solve. This is the price to pay at the end of the day.

3.6.1 Radiative Transfer in Disks

Because of the huge numerical efforts required to solve the coupled equations of SE and RT, the problem is often approximately split between continuum and line radiative transfer. If the line emission has negligible effects on the overall radiation field, this approximation is not so bad. The SE equations are then solved by using a simplified (sometimes multi-directional) escape probability approach, where the local radiation field $P(v_{ij})$, which is used in the SE equations, is replaced by a mixture of the “background” continuum radiation field J_v^{cont} (obtained from solving the continuum RT only) and the local line source function $B_v(T_{\text{exc}})$. The radiative transfer of line photons is treated approximately, using the column densities along one or more directions, to evaluate the probability of a local line photon to escape the medium and/or the probability of continuum photons to penetrate into the local

line “resonance” region, pumping the levels (Avrett & Hummer 1965). In 1D plane-parallel geometry, the intensity at the surface I_v is related to the intensity at the bottom $I_v(0)$ through

$$I_v = I_v(0) e^{-\tau} + B_v(T_{\text{exc}}) (1 - e^{-\tau}) . \quad (45)$$

Roughly speaking, $e^{-\tau}$ gets replaced by the escape probability $\beta(\tau)$, where τ is the line optical depths along the considered ray direction. This approximation has problems if the geometry is multi-dimensional (no plane-parallel geometry) and the medium becomes optically thin. It will be discussed in more detail in the chapter by Woitke (2015). Finally, the line emission is calculated by performing ray tracing through the computational volume and using the level populations derived from the escape probability approach.

3.6.2 Velocity gradients

A protoplanetary disk has a Keplerian velocity field. This leads to double peaked velocity profiles as described in the seminal paper of Beckwith & Sargent (1993) and detailed in the chapter of Dionatos (2015). If a line photon is emitted in cell r_n , it is shifted according to the Keplerian velocity of that cell. If the photon travels a distance $s = r_n - r_{(n-1)}$ from one cell in the midplane to the next, this corresponds to a difference in velocities $v(r_n) - v(r_{n-1})$, and hence to a frequency shift of

$$\Delta v_{ij} = v_{ij} \frac{v(r_n) - v(r_{n-1})}{c} . \quad (46)$$

If the frequency shift is larger than the line width, the photon is ‘shifted out of the line’ and can escape. In Interstellar Medium physics, this is often referred to as the Large Velocity Gradient (LVG) approximation. It resembles the escape probability approximation discussed above since the escape probability can be written as

$$\beta_{\text{esc}} = \frac{1 - e^{-\tau_{\text{LVG}}}}{\tau_{\text{LVG}}} , \quad (47)$$

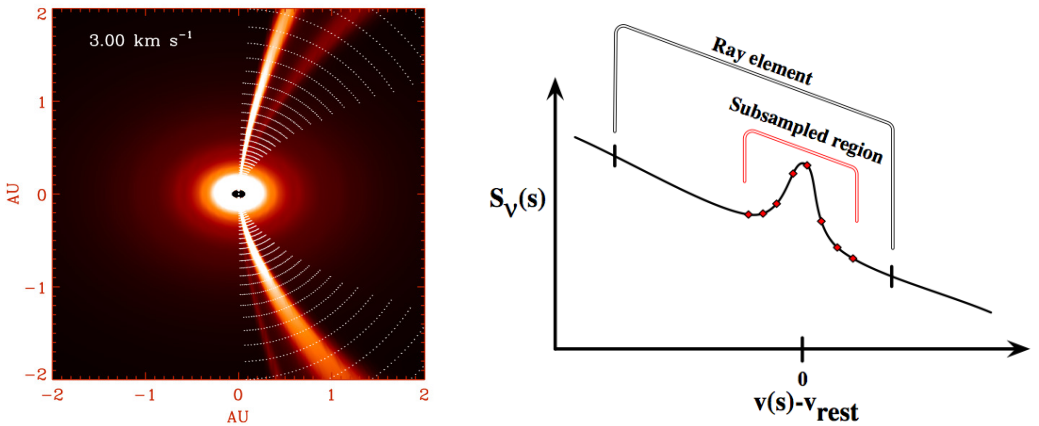


Figure 5. Left panel: Channel map of a line that is red-shifted by 3 km/s in a disk — visible are the front and back, but also the lower and upper disk part. Right panel: Sketch illustrating the concept of subsampling in the case of large velocity gradients and a finite cell size. Figures from Pontoppidan et al. (2009, © AAS, reproduced with permission).

where τ_{LVG} is the total optical depth along the path for any frequency. In disks, however, the LVG approximation does not work since a line photon can interact with the gas at various locations along a ray (e.g. top and bottom of the disk, near-side and far-side, Fig. 5, left panel). This problem is discussed in Pontoppidan et al. (2009) in detail for near-IR lines originating in the inner disk where velocity gradients across a cell can be large. In that case, the opacity needs to be properly sampled despite a finite cell size. To illustrate this, consider a typical turbulent/thermal broadening of <1 km/s and a grid with a resolution of 0.01 AU at a distance of 1 AU from the star. The velocity gradient across one cell is then ~ 4 km/s for a solar mass star. One possibility to ‘catch’ the line within the cell is a subsampling of the source function between the cell boundaries (Fig. 5, right panel).

4 Special cases

After having discussed the foundations of the statistical equilibrium and radiative transfer, we will turn to a few special cases to illustrate the basic concepts.

4.1 Radiative pumping

The molecules in the disk surface are directly exposed to the stellar radiation field. This means that besides their statistical equilibrium being governed by the warm gas temperatures in the surface, radiative pumping can be very strong if the molecule has strong transitions in the UV/optical wavelength range, where the stellar radiation field peaks. An example is the CO molecule, which has electronic bands in the UV (4th positive system: $A^1\Pi - X^1\Sigma^+$ around 1600 Å). Figure 6 illustrates the UV pumping of the upper electronic state and the cascade leading eventually to the emission of the CO ro-vibrational lines from the ground electronic state in the near-IR ($\sim 4.6 \mu\text{m}$).

The stellar radiation field enters into the equations of SE through the absorption term $\sum_{j<i} n_j B_{ji} P(v_{ji})$. The UV fluorescence can pump the populations in the upper vibrational levels of the ground electronic state to values much higher than LTE. Hence, the band ratios between the $v=1-0$, $2-1$, $3-2$ bands will be affected and the hotter bands ($v=2-1$, $3-2$) will get stronger with respect to the fundamental band ($v=1-0$). The mechanism counterbalancing this fluorescence is collisional

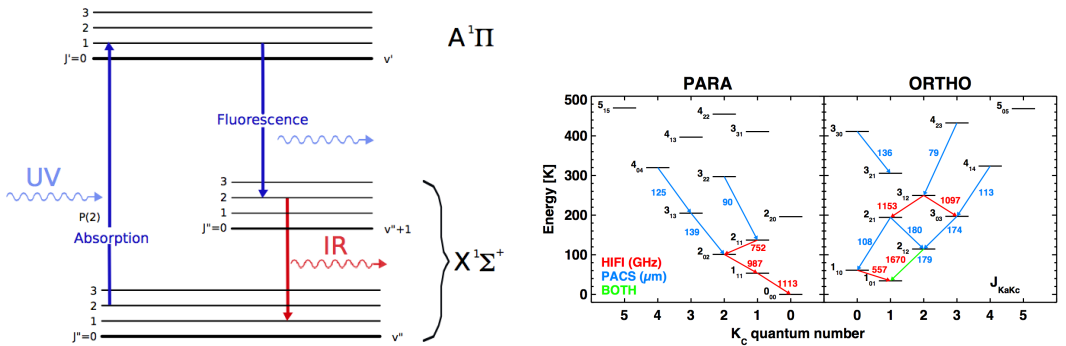


Figure 6. Left panel: Sketch of the CO fluorescence mechanism. Right panel: Sketch of the ortho- and para-water energy levels together with some key lines that have been observed with the Herschel satellite. Figures are from Thi et al. (2013, reproduced with permission © ESO), van Dishoeck et al. (2011).

quenching (i.e. collisional de-population of the higher v -levels) due to high densities. The main collision partner in the emitting region is atomic hydrogen (Thi et al. 2013) whose collision rates are two orders of magnitude larger than those of molecular hydrogen.

Another example of radiative pumping is the water molecule. It has key transitions in the mid-IR where the spectral energy distribution of protoplanetary disks peaks. This means that the IR continuum photons (e.g. at 79 and 90 μm) can efficiently pump low rotational levels of water. We saw in Sect. 2.1 that we can define an excitation temperature T_{ex} for any two levels. In the most extreme case of radiative pumping, the excitation temperature of the line would approach the local dust temperature (optically thick case) or more precise the radiation temperature of the continuum. This has been illustrated in the study of the water statistical equilibrium and radiative transfer in the disk around TW Hya (Kamp et al. 2013).

4.2 Masers

Maser (microwave amplification by the stimulated emission of radiation) emission is frequently observed in star forming regions and is often associated with massive star formation and H II regions. Maser emission is also seen in the vicinity of AGB stars. Table 1 gives an overview of detected maser transitions from molecules and references to the literature.

In order to observe a maser, we need an inversion of the level populations, so a collisional and/or radiative pumping mechanism. For the case of dusty environments around AGB stars, some possible pumping mechanisms for the 22 GHz water maser are described by Babkovskaia & Poutanen (2006). Characteristics of maser emission are large intensities, very narrow line width and unusual line ratios. The latter indicates very extreme non-LTE conditions. If multiple maser lines of the same molecule are observed from the same physical region, the excitation mechanism and the small scale structure and physical conditions in the gas can be disentangled.

Table 1. Examples of Maser transitions found in star forming environments and the vicinity of AGB stars.

Molecule	Frequency (GHz)	Typical transition	example reference
OH	1.665	$^2\Pi_{3/2}, J=3/2, F=1-1$	Asanok et al. (2010)
	1.667	$^2\Pi_{3/2}, J=3/2, F=2-2$	Asanok et al. (2010)
	1.720	$^2\Pi_{3/2}, J=3/2, F=2-1$	Asanok et al. (2010)
H ₂ CO	4.829	$1_{10}-1_{11}$	Araya et al. (2006)
CH ₃ OH	44.069	$7_0-6_1 A^+$	Fontani et al. (2010)
H ₂ O	22.235	$6_{16}-5_{23}$	Asanok et al. (2010)
NH ₃	23.870	$3,3-3,3$	Hindson et al. (2010)
SiO	43.122	$v=1, J=1-0$	Morita et al. (1992)

4.3 Resonance scattering

An important example for resonance line scattering in disks is Ly α (Bergin et al. 2003; Neufeld 1990). Figure 7 shows the escape fraction of a Ly α photon from a one dimensional slab model as a function of the atomic hydrogen column density $N(\text{H}^0)$. The two processes of dust absorption and collisional de-excitation compete in destroying Ly α photons. The solid curves show an increasing ratio of the scattering coefficient for the line photons and the dust absorption coefficient. With less efficient dust absorption, a higher fraction of line photons can escape the slab. The dashed curves illustrate the effect of decreasing collisional de-excitation. The main collisional partners are H⁺ and electrons: $\text{H}(2p) + \text{H}^+ \rightarrow \text{H}(2s) + \text{H}^+$ and $\text{H}(2p) + e \rightarrow \text{H}(2s) + e$.

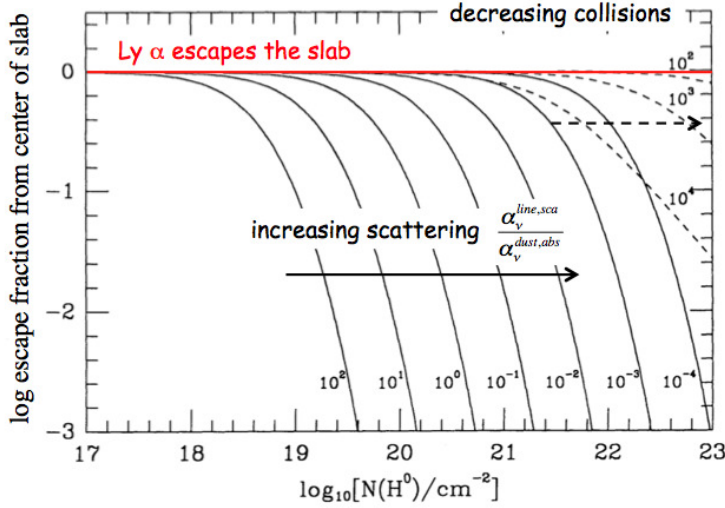


Figure 7. Escape fraction of a Ly α photon from a 1D slab model. Figure adapted from Neufeld (1990, © AAS, reproduced with permission).

Resonance scattering of Ly α photons into the disk can increase the depth into the disk to which molecules can be photo dissociated. This occurs e.g. for molecules that have their dissociation bands overlapping with Ly α such as CO, OH, H₂O and HCN (Fogel et al. 2011). Other molecules such as CN are unaffected. However, part of this can be compensated by an increased photodesorption rate also due to the enhanced Ly α radiation.

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